

Name of the Student: _____

Max. Marks : 20 Marks

Time : 20 Minutes

Mark Schemes

Q1.

- (a) Any **two** from: ✓✓
- using electrons gives greater resolution (as the wavelength can be made very small)
 - electrons can get closer to the nuclei (as there is no electrostatic repulsion)
 - electrons have less recoil (as their mass is small compared to the nucleus)
 - free electrons are easier to accelerate **OR** give energy to (as charge-to-mass ratio is higher)
 - electrons are easier to produce
 - scattering distributions are easier to interpret **OR** strong nuclear interaction is not involved
 - using alpha particles only gives the distance of closest approach/upper limit to the radius.

OWTTE on each advantage

Allow reverse arguments

2

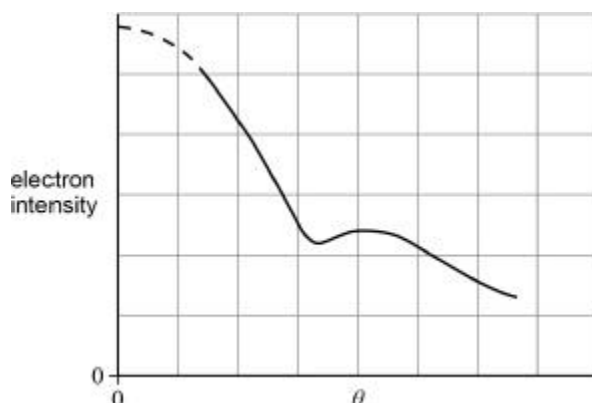
- (b) A curved line showing an decrease in intensity with increase in θ . ✓

There is a single **non-zero** minimum. ✓

Do not allow U-shaped graphs.

The initial part of the curve may be absent.

Award MAX 1 for a line that covers less than half the θ axis.



2

- (c) Density (= mass÷volume) = $\frac{Am_{\text{nucleon}}}{\frac{4}{3}\pi R^3}$ ✓₁

$$\left(\text{substituting density} = \frac{Am_{\text{nucleon}}}{\frac{4}{3}\pi(R_0 A^{\frac{1}{3}})^3} \right)$$

✓₁ Do not accept M unlabelled.

Accept m (ie lowercase) unlabelled.

Condone m_n (mass of neutron in Data sheet) for m_{nucleon} .

Accept only 1.67×10^{-27} (kg) for mass of nucleon (ie to 3 sf).

$$\text{Density} = \frac{3m_{\text{nucleon}}}{4\pi(R_0)^3} \quad (\text{in which) all (terms are) constant OR the expression does not depend on } A \quad \checkmark_2$$

Allow ecf to MP2 for any misrepresentation of m provided it is clear that it signifies mass of a single nucleon.

✓₂ If the constants are identified the equation may be converted into a ratio. The equation may be rearranged to have R_0 as the subject with work to show that this is a constant when the density is constant.

2

(d) Any **one** of: ✓

The mass of the nucleus is not exactly $A \times m_{\text{nucleon}}$

(because this ignores the binding energy)

OR

The volume equation assumes that the nucleus is a perfect sphere (which is not true) OWTTE

OR

The density equation that uses the nuclear radius formula implies that the density is uniform within a nucleus which is not true. OWTTE

OR

Protons have a slightly different mass to a neutron

Do not accept "density of individual nucleons can be different from each other". It is not allowed as it does not occur in the working equation.

1

$$(e) \quad R_0 = \left(\frac{R}{A^{\frac{1}{3}}} = \frac{4.02 \times 10^{-15}}{(35)^{\frac{1}{3}}} \right) = 1.2(3) \times 10^{-15}(\text{m}) \quad \checkmark_1$$

Substitutes values into density equation ✓₂

$$\checkmark_2 \text{ e.g. Density} = \left(\frac{\text{mass}}{\text{volume}} = \frac{Am}{\frac{4}{3}\pi R^3} \right) = \frac{35 \times 1.67 \times 10^{-27}}{\frac{4}{3}\pi(4.02 \times 10^{-15})^3}$$

$$\text{Density} = 2.1 \times 10^{17} (\text{kg m}^{-3}) \quad \checkmark_3$$

✓₃ Evidence of a calculation must be given to gain this mark

Accept 2.15 but not 2.2

3

Q2.

(a) Heavy water

OR

Beryllium / Be

OR

(normal) Water ✓

Accept D₂O and H₂O

1

(b) Any **two** points from: ✓✓

- U-235/Uranium fuel will (be more likely to) absorb the neutron
 - slow neutrons are less damaging **OR** cause less fatigue to the structure of the reactor/shielding/etc
 - slow neutrons (spend longer within the fissionable material and) increase the chance of causing fission
 - slowing neutrons transfers heat energy to the moderator (which can make heat easier to extract)
- all points OWTTE

*Condone the answer:**As an alternative to the first point Fission of U-236 is much more likely.**Condone the answer:**Absorption by U-238 is less likely.*

2

(c)
$$\frac{\text{final kinetic energy}}{\text{initial kinetic energy}} = \left(\frac{\frac{1}{2} m_N v^2}{\frac{1}{2} m_N u^2} \right) = \left(\frac{v}{u} \right)^2 = 0.85^2 = 72\% \quad \checkmark_1$$

(Hence) proportion of kinetic energy lost = 28% ✓₂*✓₁ can be for any of the terms shown equating to the kinetic energy ratio.**✓₂ can be an ecf but only for an arithmetic error.*

2

(d) final kinetic energy = $\left(\frac{3}{2} kT \right) = 7.2 \times 10^{-21} \text{ (J)} \quad \checkmark_1$

initial kinetic energy = $(W = QV) = 1.6 \times 10^{-13} \text{ (J)} \quad \checkmark_2$

OR

final kinetic energy = 0.045 (eV) ✓₁

initial kinetic energy = $1.0 \times 10^6 \text{ (eV)} \quad \checkmark_2$

(Rearranging equation $y = \frac{\ln\left(\frac{E_0}{E_r}\right)}{0.73}$)

✓₁₊₂ Both marks must come from the same alternative route and have consistent units (which may not be seen).

✓₁ Initial kinetic energy =

$$\frac{3}{2} \times 1.38 \times 10^{-23} \times 350 = 7.245 \times 10^{-21} \text{ J}$$

$$= \frac{7.245 \times 10^{-21}}{1.60 \times 10^{-19}} = 0.045 \text{ eV}$$

✓₂ Using the eV unit alternative the second mark cannot be given without an attempt at the first mark.

The 1.0×10^6 eV can be seen in a later substitution provided eV is used throughout.

$y = 23.(2)$ ✓₃

$$\checkmark_3 y = \frac{\ln\left(\frac{1.0 \times 10^6}{0.045}\right)}{0.73} = 23.2$$

Condone answer 24 provided it is given as an integer.

3

- (e) Idea that the model/**Figure 3** shows that low nucleon number (and so low mass) gives a greater change/reduction in speed/KE (in a collision) ✓

Idea that fewer collisions needed (with a low mass number so moderator can be thinner) ✓

Condone the use of nuclear mass instead of mass number.

2

[10]