

Name of the Student: _____

Max. Marks : 27 Marks

Time : 27 Minutes

Mark Schemes

Q1.

$$(a) \quad (i) \quad \lambda = \frac{0.693}{t_{1/2}} = \frac{0.693}{3420 \times 365 \times 24 \times 3600} \quad (1)$$

$$= 6.43 \times 10^{-12} \text{ (s}^{-1}\text{)} \quad (1)$$

$$(ii) \quad N = \frac{1}{\lambda} \frac{dN}{dt} = \frac{450 \times 10^3}{6.43 \times 10^{-12}} \quad (1)$$

$$= 7.0 \times 10^{16} \quad (1)$$

(4)

(b) (i) pass through with no [or very small] deflection (1)

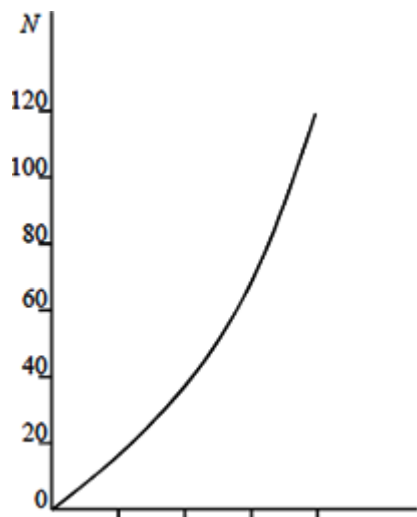
(ii) volume of nucleus \ll volume of atom (*)
 [or nucleus small and atom mostly empty space] (*)
 most of mass in nucleus (*)
 nucleus has positive charge (*)
 size of nucleus \ll separation (*)
 (*) any two (1) (1)

(3)

[7]

Q2.

(a)



straight line between $(Z = 0, N = 0)$ to $(Z = 20, N = 20)$ **(1)**
 curving upwards to $Z = 80; N = 110 - 130$ **(1)**

(2)

- (b) (i) A = any region below the line of stability
 but $N > 80$ and $Z > 60$
- (ii) B = any region above and close to the line of stability **(1)**
- (iii) C = any region below and close to the line of stability **(1)**

(3)

(c)

mode of decay	change in proton number, Z	change in neutron number, N
α emission	-2	-2
β^- emission	+1	-1
β^+ emission	-1	+1
e capture	-1	+1
p emission	-1	0
n emission	0	-1

(1)(1) (1) – lose one mark for each row in error

(3)

[8]

Q3.

(a) $(R^3 = R_0^3 A)$

plot R^3 against A with axes labelled **(1)**

units on axes **(1)**
 scales chosen to use more than 50% of page **(1)**

element	$R/10^{-15} \text{ m}$	A	$R^3/10^{-45} \text{ m}^3$
carbon	2.66	12	18.8
silicon	3.43	28	40.4
iron	4.35	56	82.3
tin	5.49	120	165.5
lead	6.66	208	295

calculate data for table **(1)**
 plot data **(1)(1)** lose one mark for each error
 calculation of gradient

e.g. gradient = $\frac{1}{3}$ **(1)** ($= 1.41 \times 10^{-45} \text{ m}^3$)

r_0 (= gradient)^{1/3} **(1)**

$= (1.41 \times 10^{-45})^{1/3} = 1.1(2) \times 10^{-15} \text{ m}$ **(1)**

alternative:

plot R against $A^{1/3}$ with axes labelled **(1)**
 units on axes **(1)**
 scales chosen to use more than 50% of page **(1)**

element	$R/10^{-15} \text{ m}$	A	$A^{1/3}$
carbon	2.66	12	2.29
silicon	3.43	28	3.04
iron	4.35	56	3.83
tin	5.49	120	4.93
lead	6.66	208	5.93

calculate data for table **(1)**
 plot data **(1)(1)** lose one mark for each error
 calculation of gradient

e.g. gradient = $\frac{6.72 \times 10^{-15}}{6.0}$ **(1)** ($= (1.1(2) \times 10^{-45} \text{ m}^3)$)

r_0 = gradient **(1)**

$= 1.1(2) \times 10^{-15} \text{ m}$ **(1)**

[or plot $\ln R$ against $\ln A$...]

(max 8)

- (b) assuming the nucleus is spherical
ignoring the gaps between nucleons
assuming all nuclei have same density
assuming total mass is equal to mass of constituent nucleus
any one assumption **(1)**

$$M = \frac{4}{3} \pi R^3 \rho \quad \mathbf{(1)}$$

$$\left(\therefore M = \frac{4}{3} \pi R_0^3 a \rho \right)$$

$$\left(\therefore \rho = \frac{3m}{4\pi R_0^3} \right) = \frac{3 \times 1.67 \times 10^{-27}}{4\pi \times (1.12 \times 10^{-15})^3} \quad \mathbf{(1)}$$

$$= 2.8 \times 10^{17} \text{ kg m}^{-3} \quad \mathbf{(1)}$$

(4)

[12]