

Student: _____

Max. Marks : 20 Marks

Time : 20 Minutes

Q1.

The photograph shows the containers of two radioactive sources kept in a school.



The isotope Ra 226 undergoes a series of decays until it produces the stable isotope Pb 206.

Determine the number of α particles and β particles emitted during this process to complete the nuclear equation.

(3)

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Q2.

Tomatoes can be made into a puree.

The tomato puree has a mass of 0.444 kg and boils at 101°C. 175kJ of energy are supplied to bring it to its boiling point from a temperature of 21°C.

Determine the specific heat capacity of the puree.

(2)

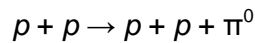
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Specific heat capacity =

(Total for question = 2 marks)

Q3.

A high energy proton collides with a stationary proton and a π^0 particle is produced.
The equation for the reaction is



(i) Explain why the proton must have a high energy in order for this reaction to occur.

(2)

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(ii) The rest mass of the π^0 is $\frac{1}{7}$ of the rest mass of a proton.
In this reaction the total kinetic energy of the particles decreases.
Calculate the minimum decrease in kinetic energy if the reaction is to occur.
rest mass of proton = 938 GeV/c²

(2)

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Minimum decrease in kinetic energy =

Q4.

Phosphogypsum is a by-product in the manufacture of fertiliser. It is slightly radioactive because of the presence of radium-226, a radioisotope with a half-life of 1600 years.

It must be stored securely as long as the activity of the radium-226 it contains is greater than 0.4 Bq per gram of phosphogypsum.

Radium-226 decays to radon-222 by alpha emission.

Determine the energy released in MeV in the decay of a single nucleus of radium-226.

(5)

mass of radium-226 nucleus = 225.97713 u

mass of radon-222 nucleus = 221.97040 u

mass of α particle = 4.00151 u

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Energy released = MeV

(Total for question = 5 marks)

Q5.

A series of experiments was carried out in the 1970s to investigate the structure of protons using the linac at Stanford, USA.

* Explain how an electron is accelerated in a linac.

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(Total for question = 6 marks)