

Name of the Student: _____

Max. Marks : 17 Marks

Time : 17 Minutes

Q1.

(a) Alpha particles (α), beta particles (β) and gamma rays (γ) are types of nuclear radiation.

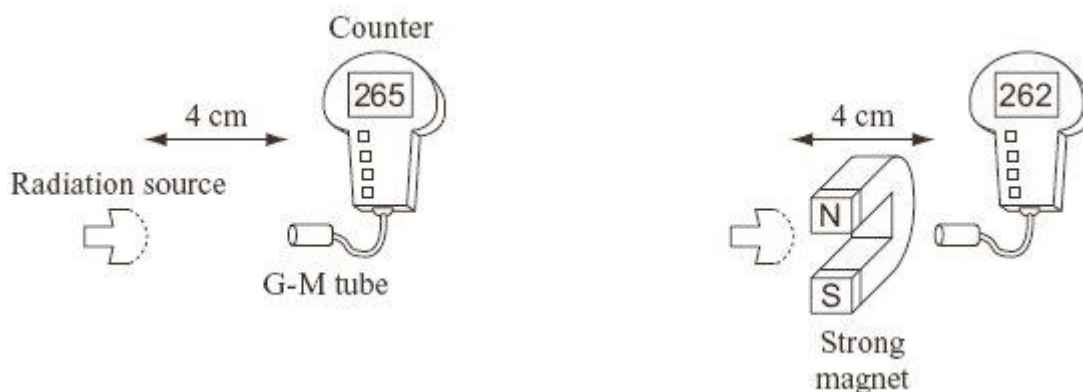
(i) Which of the three types of radiation is the most strongly ionising?

(1)

(ii) What effect does nuclear radiation have on living cells?

(1)

(b) The diagrams show a G-M tube and counter used to measure the radiation emitted from a source. Both diagrams show the reading on the counter one minute after it was switched on.



Explain why the counter readings show that the source is giving out only gamma radiation.

(2)

(c) The box gives information about the radioactive isotope technetium-99.

Type of radiation emitted: gamma
Half-life: 6 hours

Used as a medical tracer

What is meant by the term *half-life*?

(1)

- (d) To study the blood flow in a patient's lungs, a doctor injects a small quantity of a technetium-99 compound into the patient. The radiation emitted by the technetium-99 atoms is detected outside the patient's body.

Explain why a doctor would not use a radioactive isotope with a very short half-life, such as 2 seconds, as a medical tracer.

(2)

(Total 7 marks)

Q2.

- (a) Nuclear power stations use the energy released by *nuclear fission* to generate electricity.

- (i) Explain what is meant by *nuclear fission*.

(2)

- (ii) How does nuclear fission lead to a chain reaction?

You may give your answer as a labelled diagram.

(1)

- (b) Although nuclear fuels are relatively cheap the total cost of generating electricity using nuclear fuels is expensive. Why?

(1)

(c) The table compares the energy released from 1 kg of coal and 1 kg of uranium.

Coal	29 MJ
Uranium	580 000 MJ

1 MJ = 1 000 000 joules

State **one** benefit to the environment of using a concentrated fuel like uranium to generate electricity rather than using the energy from coal.

(1)

(Total 5 marks)

Q3.

A beta particle is a high-energy electron.

(i) Which part of an atom emits a beta particle?

(1)

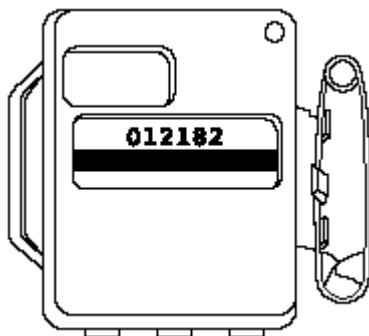
(ii) How does the composition of an atom change when it emits a beta particle?

(1)

(Total 2 marks)

Q4.

The diagram shows a badge used to monitor radiation. It measures the amount of radiation a worker has been exposed to in one month.



(i) What is used inside the badge to detect radiation?

(1)

(ii) What would indicate that the worker has been exposed to a high level of radiation as opposed

to a low level of radiation?

(1)

(iii) Why is it important to monitor the amount of radiation the worker has been exposed to?

(1)

(Total 3 marks)