

Name of the Student: _____

Max. Marks : 19 Marks

Time : 19 Minutes

Q1.

An atom of calcium, ${}_{20}^{48}\text{Ca}$, is ionised by removing two electrons.

- (i) State the number of protons, neutrons and electrons in the ion formed.

protons _____

neutrons _____

electrons _____

(3)

- (ii) Calculate the charge of the ion.

charge _____ C

(1)

- (iii) Calculate the specific charge of the ion.

specific charge _____ C kg^{-1}

(2)

(Total 6 marks)

Q2.

When ultraviolet light of frequency $3.0 \times 10^{15} \text{ Hz}$ is incident on the surface of a metal, electrons of maximum kinetic energy $1.7 \times 10^{-18} \text{ J}$ are emitted.

- (a) Explain why the emitted electrons have a range of kinetic energies up to a maximum value.

(3)

- (b) (i) Show that the work function of the metal is 1.8 eV.

(3)

- (ii) Calculate the threshold frequency of the metal. Give your answer to an appropriate number of significant figures.

threshold frequency _____ Hz

(3)

- (c) (i) State and explain the effect on the emitted electrons of decreasing the frequency of the incident radiation whilst keeping the intensity constant.

(2)

- (ii) State and explain the effect on the emitted electrons of doubling the intensity of the incident radiation whilst keeping the frequency constant.

(2)
(Total 13 marks)