

Name of the Student: \_\_\_\_\_

Max. Marks : 26 Marks

Time : 26 Minutes

Mark Schemes

Q1.

Question Number	Answer		Mark
(a)	Use of $Q = CV$ $Q = 0.18 \text{ C}$  <u>Example of calculation</u> $Q = 150 \times 10^{-6} \text{ F} \times 1200 \text{ V}$ $Q = 0.18 \text{ C}$	(1) (1)	2
(b)	Use of $W = \frac{1}{2} CV^2$ Or of $W = \frac{1}{2} QV$ Or of $W = \frac{1}{2} Q^2/C$ $W = 110 \text{ J}$ Allow ecf from (a) if $\frac{1}{2} QV$ or $\frac{1}{2} Q^2/C$ used  <u>Example of calculation</u> $W = \frac{1}{2} \times 150 \times 10^{-6} \text{ F} \times (1200 \text{ V})^2$ $W = 108 \text{ J}$	(1) (1)	2
(c)(i)	$R = 86 \text{ } (\Omega)$  <u>Example of calculation</u> $R = V/I = 1200 \text{ V} / 14 \text{ A}$ $R = 85.7 \text{ } \Omega$	(1)	1
(c)(ii)	$Q = 0.25 Q_0$ Or $Q = 0.045 \text{ C}$ Use of $RC$ (0.013 s) Use of $Q = Q_0 e^{-t/RC}$ to give $t = 0.018 \text{ s}$ (show that value will give $t = 0.019 \text{ s}$ )  [ Use of $\ln 4$ gives the correct answer if the $-$ sign is ignored , scores 1 for use of $RC$ use of $\frac{3}{4}Q \rightarrow 3.7 \times 10^{-3} \text{ s}$ scores 1 mark]  <b>Or</b> Use of $RC$ Use of $2 \times 0.69 \times RC$ $t = 0.018 \text{ s}$  <u>Example of calculation</u> $Q = 0.25 Q_0$ $Q = Q_0 e^{-t/RC}$ $0.25 Q_0 = Q_0 e^{-t/RC}$ $\ln(0.25) = -t / (86 \text{ } \Omega \times 150 \times 10^{-6} \text{ F})$ $t = 0.0178 \text{ s}$	(1) (1) (1)	3

(c)(iii)	Same charge (flows for shorter time) <b>OR</b> (Same charge flows for) shorter time	(1)	1
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Q2.

Question Number	Acceptable answers	Additional guidance	Mark
i	<ul style="list-style-type: none"> <li>Use of <math>\ln V = \ln V_0 - \frac{t}{RC}</math> (1)</li> <li>Substitution <math>V = 2.0</math> V and <math>V_0 = 8.0</math> V (1)</li> <li><math>t = 5.6(1)</math> ms (1)</li> </ul>	<p>Alternative use of <math>V = V_0 e^{-\frac{t}{RC}}</math></p> <p>Rearrange to <math>\ln 4 = t / 2700 \Omega \times 1.5 \times 10^{-6} F</math></p> <p><u>Example of calculation</u></p> <p><math>t = 2700 \Omega \times 1.5 \times 10^{-6} F (\ln 8 - \ln 2)</math> <math>t = 5.61</math> ms</p>	3
ii	<ul style="list-style-type: none"> <li>Use of <math>W = \frac{1}{2} CV^2</math> (1)</li> <li><math>W = 3.0 \times 10^{-6}</math> J (1)</li> </ul>	<p><u>Example of calculation</u></p> <p><math>W = \frac{1}{2} 1.5 \times 10^{-6} F \times 2^2 V^2 = 3.0 \times 10^{-6} J</math></p>	2

Q3.

Question Number	Answer	Mark
(a)	<p>At least three vertical lines spread over symmetrically over more than half of the plate length and touching both plates. (1)</p> <p>(ignore edge ones that might curve)</p> <p>All equispaced and parallel [don't allow gapping to avoid oil drop] (1)</p> <p>Arrow pointing downwards (1)</p>	3
(b)	<p>Negative / - / -ve (1)</p> <p>( negative and/or positive does not get the mark)</p>	1

(c)	<p>Upward force labelled: Electric (force) <b>Or</b> Electrostatic (force)  <b>Or</b> force due to electric field <b>Or</b> electromagnetic (force) (1)  [do not accept repulsive/attractive force. If EQ used, the symbols must be defined]</p> <p>Downward force labelled: mg, weight, W, gravitational force (1)</p> <p>(for both marks the lines must touch the drop and be pointing away from it. Ignore upthrust if drawn but one mark lost for each extra force added)</p>	2
(d)(i)	<p><math>E = 5100 \text{ V} / 2 \text{ cm}</math> (1)  Conversion of cm to m (1)  Use of <math>QE = mg</math> (<math>1.18 \times 10^{-13} \text{ kg}</math>) (1)  <math>Q = 4.6 \times 10^{-19} \text{ C}</math> (1)</p> <p>(<math>E = 255\,000 \text{ (V m}^{-1}\text{)}</math> scores MP1 &amp; 2.  unit conversion missed <math>\rightarrow Q = 4.62 \times 10^{-17} \text{ C}</math> scores MP1 &amp; 3  if <math>V</math> is halved <math>\rightarrow Q = 9.23 \times 10^{-19} \text{ C}</math> scores MP1 ,2 &amp; 3)</p> <p><u>Example of calculation</u>  <math>E = V/d</math>  <math>F = EQ = mg</math>  <math>Q = mg / E = mgd/V</math>  <math>Q = (1.20 \times 10^{-14} \text{ kg} \times 9.81 \text{ m s}^{-2} \times 0.02 \text{ m}) / (5100 \text{ V})</math>  <math>Q = 4.62 \times 10^{-19} \text{ C}</math></p>	4
(d)(ii)	<p>Answer to (d)(i) divided by <math>e</math> (1)  3 electrons <b>Or</b> sensible integer number less than 500 (1)  (answers with very large numbers of electrons can get MP1 only)</p> <p><u>Example of calculation</u>  Number of electrons = <math>4.62 \times 10^{-19} \text{ C} / 1.6 \times 10^{-19} \text{ C}</math>  Number = 2.9 i.e. 3 electrons.</p>	2
Total for question		12