Practice Question Set For GCSE

**Subject: Physics** 

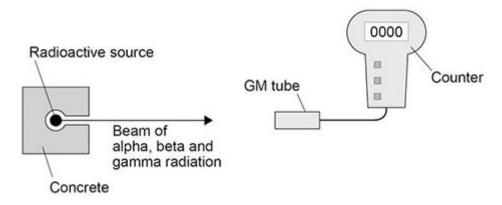


Paper-1 Topic: GCSE Triple Science Atomic Structure (Low Demand Questions)

	f the Student: orks : 20 Marks	Time : 20 Minutes
Q1.		
	dioactive isotopes emit different types of nuclear ra-	diation.
(a)	What does an alpha particle consist of?	
	Tick (✔) one box.	
	2 protons and 2 electrons	
	2 protons and 2 neutrons	
	4 protons	
	4 neutrons	(1)
(b)	What is a beta particle?	(.)
	Tick (✓) one box.	
	An electron	
	A neutron	
	Electromagnetic radiation	
		(1)
(c)	A krypton (Kr) nucleus decays into a rubidium (R	b) nucleus by emitting a beta particle.
	What is the correct equation for this decay?	
	Tick (✓) one box.	

(d) The figure below shows an experiment to demonstrate how alpha, beta and gamma radiation penetrate different materials.

The experiment takes place in a vacuum.



Three different materials are used:

- a sheet of paper
- a 0.5 cm thick sheet of aluminium
- a 10 cm block of lead.

Each material is placed one at a time between the radioactive source and the GM tube.

The GM tube and counter show whether the material has stopped the radiation.

Complete below table to show how alpha, beta and gamma radiation penetrate the materials in the figure above.

Use the words **Yes** and **No**.

Part of below table has been completed for you.

Type of	Most radiation is stopped by:			
Type of radiation	the sheet of paper	the sheet of aluminium	the block of lead	
Alpha			Yes	
Beta	No			
Gamma		No		

(3)

(1)

(e) Alpha, beta and gamma radiation have different ionising powers.

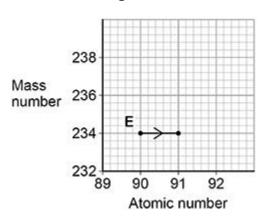
Draw **one** line from each radiation type to the correct ionising power. Radiation type lonising power Zero Alpha Low Beta Medium Gamma High (3) (f) Some sources of background radiation are natural and other sources are man-made. Which of the following is a man-made source of background radiation? Tick (✓) one box. Cosmic rays Nuclear accidents **Rocks** (1) (g) The average background radiation dose per year in the UK is 2.0 millisieverts. A dental X-ray gives a patient a radiation dose of 0.005 millisieverts. Calculate how many dental X-rays would be the same as the average background radiation dose per year. Number of dental X-rays = \_\_\_\_ (2) (Total 12 marks) **Q2**. Atoms of different elements have different properties. (a) Which of the following is the same for all atoms of the same element? Tick (✓) one box.

	Atomic number		
	Mass number		
	Neutron number		
(b)	Which of the following is differe	ent for isotopes of the same element?	(1)
	Tick (✓) one box.		
	Number of electrons		
	Number of neutrons		
	Number of protons		
			(1)
(c)	A nucleus emits radiation.		
	Figure 1 shows how the mass	number and the atomic number change.	
	The nucleus is labelled <b>D</b> .		
		Figure 1	
	Mass	228- 226- D	
	number	224	
		222 85 86 87 88 89 Atomic number	
	Which type of radiation is emitt		
	Tick (✔) one box.		
	Alpha		
	Beta		

(d) Nucleus **E** also emits radiation.

Figure 2 shows how the mass number and the atomic number change for nucleus E.

Figure 2



Which type of radiation is emitted when nucleus **E** decays?

Tick (✓) one box.

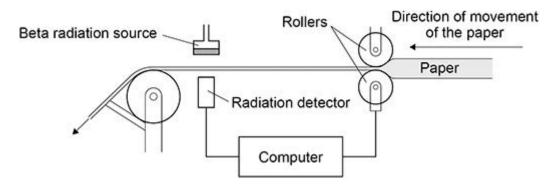
Alpha	
Beta	
Neutron	0

(1)

Beta radiation can be used to monitor the thickness of paper during production.

Figure 3 shows how the radiation is used.

Figure 3



The computer uses information from the radiation detector to change the size of the gap between the rollers.

(e) Complete the sentences.

The thickness of the paper between the beta source and the detector increases.  The reading on the detector will  This is because the amount of radiation absorbed by the paper will  All radioactive elements have a half-life.  What is meant by 'half-life'?  Tick ( / ) one box.  The time it takes for all the nuclei in a radioactive sample to split in half.  The time it takes for the count rate of a radioactive sample to halve.  The time it takes for the radiation to travel half of its range in air.  Why should the radiation source used in Figure 3 have a long half-life?  Tick ( / ) one box.  So the activity of the source is approximately constant.  So the amount of radiation decreases quickly.  So the radiation has a long range in air.	decrease	stay the same	increase	
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				(Total 8 mar

Choose answers from the box.