

Name of the Student: \_\_\_\_\_

Max. Marks : 21 Marks

Time : 21 Minutes

Mark Schemes

Q1.

Question number	Answer	Additional guidance	Mark
(i) <b>CS2</b>	An explanation to include;  there is no aluminium to absorb $\beta$ particles (1)  (therefore) more $\beta$ particles reach the G-M tube (1)	aluminium absorbs/stops/blocks beta particles  accept reverse arguments  accept radiation for beta particles	(2) <b>AO2</b>

Question number	Answer	Additional guidance	Mark
(ii) <b>CS2</b>	(idea of) background radiation	a named source of background radiation	(1) <b>AO3</b>

Question number	Answer	Additional guidance	Mark
(iii) <b>CS2</b>	becquerel	accept Bq accept close spelling	(1) <b>AO1</b>

Q2.

Question number	Answer	Mark
	<p>one from</p> <ul style="list-style-type: none"><li>• same atomic number (1)</li><li>• same number of protons (1)</li><li>• same element (1)</li></ul> <p>and one from</p> <ul style="list-style-type: none"><li>• different numbers of neutrons (1)</li><li>• different mass numbers (1)</li></ul>	<b>(2)</b>

Q3.

	Answer	Mark
(i)	${}_{95}^{245}\text{Am}$  <b>A, B and D</b> are incorrect as these are not isotopes of americium.	(1) AO1

	Answer	Mark
(ii)	<b>[x] B</b> 5 cm  <b>A, C and D</b> are incorrect as these are not the correct range of an alpha particle in air.	(1) AO1

	Answer	Additional guidance	Mark
(iii)	${}_{2}^{4}\alpha$ (1)  ${}_{115}^{237}\text{At}$ (1)  ${}_{93}^{237}\text{Np}$ (1)	both correct for the mark  ecf from mp1  ecf from mp1	(3) AO2

Q4.

Question number	Answer	Mark
	C a helium nucleus	(1)

Q5.

Question number	Answer	Mark
	B $10^{-10}$ m	(1)

Q6.

Question	Answer	Mark
	D +2  A , B and C do not apply to an alpha particle	<b>1</b> <b>AO1.1</b>

Q7.

Question	Answer	Mark
	<b>D</b> it requires high temperature and pressure <b>A</b> no emission of daughter nuclei <b>B</b> not a chain reaction <b>C</b> does not produce radioactive waste	<b>1</b> <b>AO1.1</b>

Question Number:	Answer	Mark
(i)	<p><input type="checkbox"/> C a neutron</p> <p><b>The only correct answer is C (neutron causes U-235 fission)</b></p> <p><b>A</b> is not correct – incorrect particle</p> <p><b>B</b> is not correct – incorrect particle</p> <p><b>D</b> is not correct – incorrect particle</p>	(1)

Question Number	Answer	Additional guidance	Mark
(ii)	<p>recall and substitution (1)</p> $1.2 \times 10^{-11} = \frac{1}{2} \times 1.4 \times 10^{-25} \times v^2$ <p>rearrangement (1)</p> $v^2 = \frac{2 \times 1.2 \times 10^{-11}}{1.4 \times 10^{-25}}$ <p>evaluation (1)</p> <p>(v=) <math>1.3 \times 10^7</math> (m/s)</p>	<p>accept rearrangement and substitution in either order</p> <p>ignore POT until evaluation</p> $v^2 = 1.71 \times 10^{14}$ <p>allow numbers that round to <math>1.3 \times 10^7</math> (m/s)</p> <p>1.3 to any other power of ten scores 2 marks</p> <p>award full marks for the correct answer without working</p>	(3)





Q9.

Question number	Answer	Mark
	<input checked="" type="checkbox"/> C beta plus Options A, B and D are not represented by Figure 9.	(1)

Q10.

Question number	Answer	Mark			
	<table><tr><td><b>A</b></td><td>same atomic number</td><td>different number of neutrons</td></tr></table> <p>Options B, C and D have wrong combinations</p>	<b>A</b>	same atomic number	different number of neutrons	<b>(1)</b>
<b>A</b>	same atomic number	different number of neutrons			